

83% AMMONIUM NITRATE SOLUTION

Section 1 – Identification

Product	83% Ammonium Nitrate Solution (83.0% A.N. Solution)	Recommended Use:	As a component in the manufacturing of various industrial products.
Manufacturer	TradeMark Nitrogen Corp.		
Address	1216 Old Hopewell Road, Tampa, FL 33619		
Phone	(813) 626-1181 (800) 452-3107		
24 Hour Contact	Chemtrec (800) 424-9300		

Section 2 – Hazard Identification



GHS03

Warning: Oxidizer
May intensify fire; Store away from organics or other oxidizable materials; In case of fire: use water to put out fire.



GHS07

Warning: Irritant
Wear protective gloves, protective clothing, eye & face protection, and respiratory protection.



GHS08

Warning: Irritant
May cause damage to organs (blood).

Section 3 – Composition

Ingredients	Component	CAS. No.	Percent by Weight
	Ammonium Nitrate (NH ₄ NO ₃)	6484-52-2	83.00%
	Water (H ₂ O)	7732-18-5	17.00%

Section 4 – First Aid Measures

Inhalation	If inhaled: Remove person to fresh air and keep comfortable for breathing. Provide artificial respiration if necessary. Seek medical attention.
Skin Contact	If on skin (or hair): Take off all contaminated clothing. Rinse skin with soap and water for at least 15 minutes. Liquid is hot - may need to treat exposed person for burns.
Eye Contact	If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing for at least 15 minutes. Seek medical attention if irritation persists.
Ingestion	If swallowed: Do NOT induce vomiting. Drink large amounts of water. Never give anything by mouth to an unconscious person. Seek medical attention.
Acute Health Hazards	High levels of nitrates may reduce the blood's ability to transport oxygen causing headache, fatigue, dizziness and blue lips and skin (methemoglobinemia).
Chronic Health Hazards	Methemoglobinemia is the primary health effect, but possible excessive action of the kidneys and perhaps bowels can occur.

Section 5 – Fire Fighting Measures

Suitable Extinguishing Techniques & Equipment	83% Ammonium Nitrate is non-flammable aqueous solution. Flooding quantity of water is recommended in the event of a fire. Do not use salt water, carbon dioxide, dry chemicals or foam extinguishers.
Chemical Hazards From Fire	If product evaporates, residual solid can be explosive. In a fire, carbon oxides, nitrogen oxides and ammonia may be present.
Special Fire Fighting Procedures	Keep material wet to prevent nitrate salts from forming as they can support combustion or become unstable. Avoid contamination of ammonium nitrate with organic materials such as oil, sulfur, metal fines or other combustible substances as the mixture may become unstable. For large fires, apply water to the sides of the container from a distance. If that is not
NFPA Rating	Health - 1 (Slight) Fire - 0 (Least) Reactivity - 3 (High) OX - Oxidizer
Other	Do not allow run-off from fire fighting to enter drains or water



Section 6 – Accidental Release Measure

Personal Precautions	Avoid splashing. Prevent exposure to spilled material with the use of proper PPE.
Protective Equipment	PPE should include gloves, goggles, face shield and level C protective suit.
Containment	Control the flow of product using dikes of soil, sand bags or other commercially available inert sorbent socks or booms.
In Case of Spill	Absorb product with inert absorbent. Avoid splashing or spraying. Contain and pick up spill in diked area. Prevent discharge to sewers or water ways. If uncontaminated, recover and re-use.

Section 7 – Safe Handling & Storage

Precautions for Safe Handling & Storage	Store in a well ventilated cool dry place. Avoid heating Ammonium Nitrate Solution in a confined space (i.e. pipe, pump, etc.) as the solution may decompose and explode. Avoid welding on pipes or tanks that have contained Ammonium Nitrate Solution until they have been thoroughly washed out with water.
Incompatibility	Avoid contact with readily oxidizable materials, strong acids and chlorates. Contact with alkaline materials will produce ammonia. Will corrode copper, bronze and brass.

Section 8 – Exposure Controls / Personal Protection

Exposure Limits	Component	Permissible Exposure Limit	Threshold Limit Value	Short Term Exposure Limit	Immediately Dangerous to Life or Health
	Ammonium Nitrate (NH ₄ NO ₃)	Not Established	Not Established	Not Established	Not Established
	Water (H ₂ O)	Not Established	Not Established	Not Established	Not Established
Engineering Controls	Local or general exhaust. Eyewash facilities should be available.				
Personal Protective Equipment	Eyes	Chemical safety goggles or safety glasses.			
	Hands	Impervious chemical protective gloves.			
	Respiratory	None required under normal conditions. NIOSH approved respirator if there is a mist of the product			
	Protective Clothing				



Gloves



Goggles



Protective Clothing



Respiratory Protection

Section 9 – Physical & Chemical Properties

Appearance and Odor	Clear liquid with little to no detectable odor.	Relative Density	1.370 at 175°F
Boiling Point	> 212°F at 1 atmosphere	Molecular Weight	80.05
Freezing Point	N/A	Solubility in Water	Miscible in Water
Vapor Pressure	0.06 psia at 60°F	Evaporative Rate	No Data Available
Weight per Gallon	11.43 lbs/gal @ 175°F	pH	2.0 - 6.0
Flash Point	No Data Available	Salt-Out Temp	155°F (68.3°C)
Flammability Limits	No Data Available	Auto Ignition Temp	Not Flammable
UEL	No Data Available	LEL	No Data Available

Section 10 – Stability & Reactivity

Reactivity	Product is not reactive under normal conditions.
Stability	Product is stable under normal conditions.
Hazardous Reactions	Hazardous polymerization will not occur.
Conditions to Avoid	Do not allow product to evaporate to dryness. Keep away from heat. Avoid heating within a confined space. Avoid incompatibilities, contamination and combustible materials
Incompatible Materials	Avoid contact with readily oxidizable materials, strong acids and chlorates. Contact with alkaline materials will produce ammonia. Will corrode copper, bronze and brass.
Hazardous Decomposition Products	If product evaporates, residual solid (ammonium nitrate) can be explosive. In a fire, carbon oxides, nitrogen oxides and ammonia may be present.

Section 11 – Toxicology Information

Routes of Exposure	Inhalation, ingestion or skin/eye absorption
Symptoms and Signs of Exposure	Eyes Causes eye irritation. Skin Mild irritant.
Long Term Effects	Inhalation May irritate respiratory tract causing cough and sore throat. Ingestion Can cause abdominal pain, vomiting, diarrhea and methemoglobinemia. Methemoglobinemia is the primary long-term health effect.
Toxicity	Ammonium Nitrate Rat Oral Toxicity LD ₅₀ 2217-4500 mg/kg (OECD Guide 401)
Carcinogen	The International Agency for Research on Cancer has not classified ammonium nitrate for its carcinogenic potential (IARC 1987).

Section 12 – Ecological Information

Water	Low concentrations are not toxic to fish or other aquatic organisms. High concentrations may be toxic to aquatic life and encourage excessive algae growth.
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Section 13 – Disposal Considerations

Waste	Ammonium Nitrate is not considered a hazardous waste. Disposal must be done in accordance with local, state and federal environmental regulations. Place waste in an appropriate container with correct labeling.
Additional Information	This material is highly water soluble. Landfills receiving this material should be equipped to contain leachate.

Section 14 – Transport Information

This material is hazardous as defined by 49 CFR 172.101 by the US Department of Transportation

UN ID Number	UN2426
Proper Shipping Name	Ammonium Nitrate, Liquid, (Hot Concentrated)
Hazard Class	5.1
Packing Group	III
US DOT Label	5.1 (Oxidizer), Hot
Authorized Packaging	Trucks: Stainless steel MC 307, 312, DOT 407, 4 Rail: Stainless steel DOT 103, 104, 105, 109, 111, 112, 114, 115,
ERG Number	140



Section 15 – Regulatory Information

United States - SARA Hazard Category: This product has been reviewed according to the EPA Hazard Categories promulgated under Sections 311 and 312 of Title III of the Superfund Amendments and Reauthorization Act (SARA) and is considered, under applicable definitions, to meet the following categories:

Fire - No Pressure - No Reactive - No Acute - Yes Chronic - No

SARA Title III Information: This product contains the following substances subject to the reporting requirements of Title III (EPCRA) of the Superfund Amendments and Reauthorization Act of 1986 and 40 CFR Part 372:

Chemical	CAS No.	CERCLA RQ (lb)	SARA Reporting
Ammonium Nitrate	6484-52-2	N/A	302 304 313 N/A N/A Yes ⁽¹⁾

⁽¹⁾ As nitrate compounds (water dissociable)

CERCLA / Superfund, 40 CFR Part 117, 302: If this product contains components subject to substances designated as CERCLA reportable Quantity (RQ) Substances, it will be designated in the above table with the RQ value in pounds. If there is a release of RQ Substance to the environment, notification to the National Response Center, Washington DC (800-424-8802) is required.

TSCA: Ammonium nitrate salt (Nitric Acid Ammonium Salt (1:1)) is listed on the TSCA inventory.

Section 16 – Other Information

Date of Revision	October 2017 SDS update to meet GHS Standards. August 2014 TSCA statement revised. February 2013 revision prepared in accordance with 29 CFR 1910.1200 Appendix D to meet Global Harmonization Standards.
Disclaimer	The information contained in this SDS refers only to the specific material designated and does not relate to any process or use with any other materials. This information is furnished free of charge and is based on data believed to be accurate and reliable as of the date hereof. It is intended for use by persons possessing technical knowledge at their own discretion and risk. Since actual use is beyond our control, no warranty, expressed or implied, and no liability is assumed by TradeMark Nitrogen Corp. in conjunction with the use of this information. Nothing herein is to be construed as a recommendation to infringe any patents. TradeMark Nitrogen Corp. assumes no responsibility for injury to vendee or third persons proximately caused by abnormal use of the material even if reasonable safety procedures are followed. Furthermore, vendee assumes the risk in his use of the material.